

**Properties of selected fluids at 20°C = 293K and 1 bar = 10<sup>5</sup> N/m<sup>2</sup>**

Fluid	Density $\rho$ (kg/m <sup>3</sup> )	Viscosity $\mu$ (kg/m s)	Thermal conductivity $k$ (W/m K)	Coefficient of thermal expansion $\beta$ (K <sup>-1</sup> )	Isothermal Compressibility $\kappa_T$ (m <sup>2</sup> /N)	Specific heat at constant pressure $c_p$ (J/kgK)
Helium	0.164*	1.92x10 <sup>-5</sup>	0.150	3.41x10 <sup>-3</sup> *	1.00x10 <sup>-5</sup> *	5.21x10 <sup>3</sup> *
Air	1.19*	1.98x10 <sup>-5</sup>	0.0262	3.41x10 <sup>-3</sup> *	1.00x10 <sup>-5</sup> *	1.00x10 <sup>3</sup> *
Water	1.00x10 <sup>3</sup>	1.00x10 <sup>-3</sup>	0.597	1.8x10 <sup>-4</sup>	4.6x10 <sup>-10</sup>	4.18x10 <sup>3</sup>
Glycerin (C <sub>3</sub> H <sub>8</sub> O <sub>3</sub> )	1.26x10 <sup>3</sup>	1.49	0.286	5.0x10 <sup>-4</sup>	3.7x10 <sup>-10</sup>	2.39x10 <sup>3</sup>
Mercury	1.36x10 <sup>4</sup>	1.55x10 <sup>-3</sup>	8.69	1.82x10 <sup>-4</sup>	0.40x10 <sup>-10</sup>	1.39x10 <sup>2</sup>

\* Calculated from ideal gas relationships.

The isothermal compressibility  $\kappa_T$  and the coefficient of thermal expansion  $\beta$  are defined by the equation

$$\frac{d\rho}{\rho} = \kappa_T dp - \beta dT .$$

Surface tension at a clean air-water interface at 20°C :  $\sigma = 0.073$  N/m

AAS, for 2.25, 2000